Acta Crystallographica Section E

## Structure Reports

Online
ISSN 1600-5368

Jin-Cai Wu, ${ }^{\text {a }}$ Yi-Zhi Li, ${ }^{\text {b }}$ Ning Tang ${ }^{\text {a }}$ and Min-Yu Tan ${ }^{\text {a }}$

${ }^{\text {a }}$ Department of Chemistry, Lanzhou University, Lanzhou 730000, People's Republic of China, and ${ }^{\mathbf{b}}$ Coordination Chemistry Institute, State Key Laboratory of Coordination Chemistry, Nanjing University, Nanjing 210093, People's Republic of China

Correspondence e-mail: Ilyyjz@nju.edu.cn

## Key indicators

Single-crystal X-ray study
$T=293 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.006 \AA$
Disorder in solvent or counterion
$R$ factor $=0.047$
$w R$ factor $=0.159$
Data-to-parameter ratio $=15.2$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

## [ $\mu$-10,21-Di-tert-butyl-4(S),5(S),15(S),16(S)-tetra-phenyl-3,6,14,17-tetraazatricyclo[17.3.1.18,12]-tetracosa-1(23),2,6,8,10,12(24),13,17,19,21-decaene-23,24-diolato]( $\mu$-ethylene glycol)dinickel(II) diperchlorate dihydrate

In the title compound, $\left[\mathrm{Ni}_{2}\left(\mathrm{C}_{52} \mathrm{H}_{50} \mathrm{~N}_{4} \mathrm{O}_{2}\right)\left(\mathrm{C}_{2} \mathrm{H}_{6} \mathrm{O}_{2}\right)\right]\left(\mathrm{ClO}_{4}\right)_{2} \cdot-$ $2 \mathrm{H}_{2} \mathrm{O}$, the two Ni atoms are bridged by two O atoms from the Robson macrocyclic ligand, forming a parallelogram, which has a twofold axis. The macrocyclic plane is formed from an $\mathrm{Ni}_{2} \mathrm{O}_{2}$ center and four N atoms from the ligand. Above the macrocyclic plane an ethylene glycol molecule bridges the two Ni atoms, and the coordination geometry around each Ni atom is square pyramidal.

## Comment

It is well known that dinuclear cores exist at the active sites of many metalloenzymes and play an essential role in biological systems (Furuta et al., 1999; Arimura et al., 2001). The two cavities of the Robson macrocyclic Schiff base ligand, RobsonH $\mathrm{H}_{2}$, can accommodate two metal ions, forming a planar complex [Tandon et al., 1992; RobsonH ${ }_{2}=10,21$-di-tert-butyl-4,5,15,16-tetraphenyl-3,6,14,17-tetraazatricyclo[17.3.1.18,12]-tetracosa-1(23),2,6,8,10,12(24),13,17,19,21-decaene-23,24diol]. In the present study, the chiral RobsonH $\mathrm{H}_{2}$ ligand was derived from the condensation of 2,6-diformyl-o-methylphenol and $1(S), 2(S)$-diphenyl-1,2-diaminoethane in the presence of NaOH , and the title dinuclear $\mathrm{Ni}^{\mathrm{II}}$ compound, (I), was synthesized in order to investigate its biochemical effect. In this paper, the crystal structure of (I) is reported.

(I)

The two Ni atoms are bridged by O 1 and $\mathrm{O} 1^{\mathrm{i}}$ atoms, forming a parallelogram, which has a twofold axis (Fig. 1 and Table 1). There are also four N atoms from the ligand coordinated to the two Ni atoms in a macrocyclic plane. Above the plane, two O atoms from a glycol molecule coordinate to two Ni atoms in the axial directions. The coordination geometry around each Ni atom is square pyramidal. In the basal planes of the square pyramids, the two Ni atoms are exposed. This may be the catalytic active site of the complex. The crystal structure

Received 8 May 2003
Accepted 10 June 2003
Online 24 June 2003


Figure 1
A view of the molecular structure of (I), showing $50 \%$ displacement ellipsoids for non-H atoms. Suffix $A$ corresponds to symmetry code i in Table 1.
contains $\mathrm{O}-\mathrm{H} \cdots \mathrm{O}$ intermolecular hydrogen bonds (Fig. 2 and Table 2).

## Experimental

To a solution of 2,6-diformyl-4-tert-butylphenol ( 1 mmol ) in 5 ml ethanol at room temperature, NaOH ethanol solution ( $5 \mathrm{ml}, 0.2 \mathrm{M}$ ) was added. This mixture was stirred for 1 h and a yellow precipitate appeared. 1(S),2(S)-Diphenyl-1,2-diaminoethane ( 1 mmol ) dissolved in ethanol was then added and the resulting mixture stirred at room temperature for 0.5 h . After that, a solution of $\mathrm{Ni}\left(\mathrm{ClO}_{4}\right)_{2}$ in ethanol $(5 \mathrm{ml}, 0.2 \mathrm{M})$ was added and stirring continued for another 2 h . The reaction mixture was filtered and the precipitates obtained were washed with ethanol and dried in a vacuum. Crystals of (I) were grown from an ethanol-ethylene glycol solution.

## Crystal data

$\left[\mathrm{Ni}_{2}\left(\mathrm{C}_{52} \mathrm{H}_{50} \mathrm{~N}_{4} \mathrm{O}_{2}\right)\left(\mathrm{C}_{2} \mathrm{H}_{6} \mathrm{O}_{2}\right)\right]$ -
$\left(\mathrm{ClO}_{4}\right)_{2} \cdot 2 \mathrm{H}_{2} \mathrm{O}$
$M_{r}=1177.38$
Tetragonal, $P 4_{1} 2_{1} 2$
$a=11.536$ (3) A
$c=43.015$ (3) $\AA$
$V=5724$ (2) $\AA^{3}$
$Z=4$
$Z=4$
$D_{x}=1.366 \mathrm{Mg} \mathrm{m}^{-3}$
Data collection
Siemens P4 diffractometer
$\omega$ scans
Absorption correction: $\psi$ scan
(XPREP in SHELXTL;
Bruker, 2000)
$T_{\text {min }}=0.750, T_{\text {max }}=0.799$
8048 measured reflections
5296 independent reflections
4763 reflections with $I>2 \sigma(I)$

Mo $K \alpha$ radiation
Cell parameters from 25 reflections
$\theta=2.2-16.7^{\circ}$
$\mu=0.82 \mathrm{~mm}^{-1}$
$T=293$ (2) K
Block, orange
$0.35 \times 0.29 \times 0.27 \mathrm{~mm}$

$$
\begin{aligned}
& R_{\mathrm{int}}=0.011 \\
& \theta_{\max }=26.0^{\circ} \\
& h=-9 \rightarrow 13 \\
& k=-14 \rightarrow 13 \\
& l=-52 \rightarrow 51 \\
& 3 \text { standard reflections } \\
& \text { every } 97 \text { reflections } \\
& \text { intensity decay: } 4.0 \%
\end{aligned}
$$



Figure 2
The crystal packing of (I), viewed approximately down the $c$ axis. Dashed lines indicate the hydrogen bonds.

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.047$
$w R\left(F^{2}\right)=0.159$
$S=1.09$
5296 reflections
348 parameters
H-atom parameters constrained

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{o}{ }^{2}\right)+(0.105 P)^{2}\right. \\
& +1.99 P] \\
& \begin{array}{c}
\text { where } P=\left(F_{o}{ }^{2}+2 F_{c}{ }^{2}\right) / 3
\end{array} \\
& (\Delta / \sigma)_{\max }<0.001 \\
& \Delta \rho_{\text {max }}=0.49 \mathrm{e}^{-3} \\
& \Delta \rho_{\min }=-0.58 \mathrm{e}^{-3} \\
& \text { Absolute structure: Flack (1983), } \\
& 2037 \text { Friedel pairs } \\
& \text { Flack parameter }=0.02(2)
\end{aligned}
$$

Table 1
Selected geometric parameters $\left(\AA,{ }^{\circ}\right)$.

| Ni1-O1 ${ }^{\text {i }}$ | 1.837 (3) | Ni1-N1 | 1.901 (3) |
| :---: | :---: | :---: | :---: |
| Ni1-N2 | 1.874 (3) | Ni1-O2 | 2.347 (4) |
| Ni1-O1 | 1.899 (3) | $\mathrm{Ni} 1-\mathrm{Ni}_{1}{ }^{\text {i }}$ | 2.8905 (12) |
| $\mathrm{O} 1^{\mathrm{i}}-\mathrm{Ni} 1-\mathrm{N} 2$ | 172.65 (13) | $\mathrm{O} 1-\mathrm{Ni} 1-\mathrm{O} 2$ | 88.59 (12) |
| $\mathrm{O} 1^{\mathrm{i}}-\mathrm{Ni} 1-\mathrm{O} 1$ | 78.35 (12) | $\mathrm{N} 1-\mathrm{Ni} 1-\mathrm{O} 2$ | 86.02 (14) |
| $\mathrm{N} 2-\mathrm{Ni} 1-\mathrm{O} 1$ | 95.77 (12) | $\mathrm{O} 1{ }^{\text {i }}-\mathrm{Ni} 11-\mathrm{Ni} 1{ }^{\text {i }}$ | 40.09 (8) |
| $\mathrm{O} 1^{\mathrm{i}}-\mathrm{Ni} 1-\mathrm{N} 1$ | 96.12 (12) | $\mathrm{N} 2-\mathrm{Ni} 1-\mathrm{Ni}_{1}{ }^{\text {i }}$ | 134.29 (10) |
| N2-Ni1-N1 | 90.18 (13) | $\mathrm{O} 1-\mathrm{Ni} 1-\mathrm{Ni}^{1}{ }^{\text {i }}$ | 38.53 (8) |
| O1-Ni1-N1 | 171.78 (14) | $\mathrm{N} 1-\mathrm{Ni} 1-\mathrm{Ni} 1{ }^{\text {i }}$ | 135.13 (10) |
| $\mathrm{O} 1^{\mathrm{i}}-\mathrm{Ni} 1-\mathrm{O} 2$ | 96.92 (13) | $\mathrm{O} 2-\mathrm{Ni} 1-\mathrm{Ni}^{1}{ }^{\text {i }}$ | 89.93 (8) |
| $\mathrm{N} 2-\mathrm{Ni} 1-\mathrm{O} 2$ | 87.26 (14) |  |  |

Symmetry code: (i) $y, x,-z$.

Table 2
Hydrogen-bonding geometry $\left(\AA^{\circ},{ }^{\circ}\right)$.

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| O2-H2A $\cdots$ O14 | 0.97 | 1.82 | $2.744(5)$ | 159 |
| O3-H3B $\cdots$ O4 | 0.85 | 2.37 | $3.104(8)$ | 144 |
| O4-H4C $\cdots$ O14 $4^{\mathrm{ii}}$ | 0.85 | 2.17 | $2.898(7)$ | 144 |
| O4-H4C $\cdots$ O12 $2^{\mathrm{ii}}$ | 0.85 | 2.41 | $2.956(8)$ | 123 |

Symmetry code: (ii) $1+y, x,-z$.

## metal-organic papers

There are two independent water molecules of crystallization. One, O3, lies on a twofold axis, while the other, O4, shows a positional disorder over two sites, each with $50 \%$ probability. The positions of all H atoms were fixed geometrically and distances to H atoms were set by the program. H atoms were fixed geometrically with distances to H atoms in the range $0.85-0.98 \AA$ and given $U_{\text {iso }}$ alues 1.2-1.5 times the $U_{\text {eq }}$ value of the parent atom.

Data collection: CAD-4 SDP/VAX (Enraf-Nonius, 1989); cell refinement: CAD-4 SDP/VAX; data reduction: TEXSAN (Molecular Structure Corporation, 1989); program(s) used to solve structure: SHELXTL (Bruker, 2000); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

This work was supported by the Science Foundation of Gansu Province of China (No. ZR001-A25-001-Z).

## References

Arimura, K., Ohba, M., Yokoyama, T. \& Okawa, H. (2001). Chem. Lett. pp. 1134-1135.
Bruker (2000). SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.
Enraf-Nonius (1989). CAD-4 SDP/VAX. Enraf-Nonius, Scientific Instruments Division, PO Box 483, 2600 AL Delft, The Netherlands.
Flack, H. D. (1983). Acta Cryst. A39, 876-881.
Furuta, H., Ishida, A., Miyasaka, H., Fukita, N., Ohba, M., Okawa, H. \& Koikawa, M. (1999). J. Chem. Soc. Dalton Trans. pp. 367-372.
Molecular Structure Corporation (1989). TEXSAN. Version 5.0. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.
Tandon, S. S., Thompson, L. K., Bridson, J. N., McKee. V. \& Downard, A. J. (1992). Inorg. Chem. 31, 4635-4642.

